Soil Acidity impact on Nutrients

Centra In-Service

Causes of Soil Acidity

- Excessive rainfall
- Parent material from which soils develop
- Organic matter decay
- Harvest of high yielding crops
- Nitrogen fertilizer

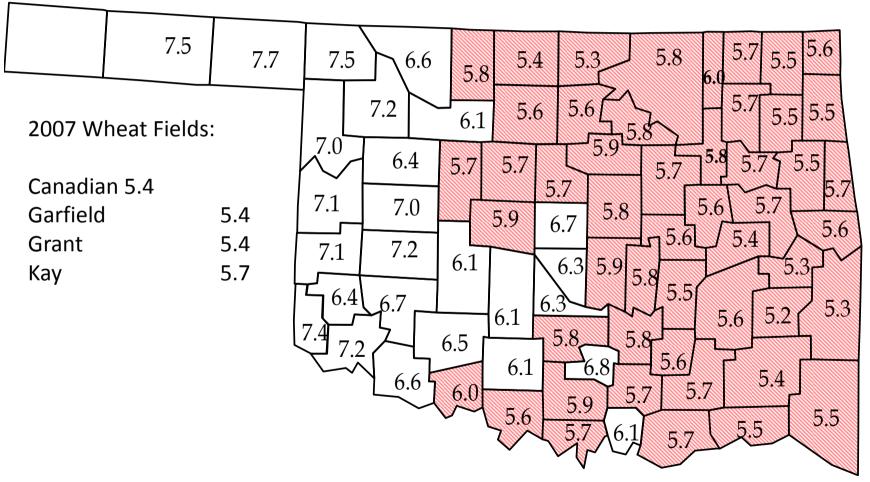
Problems with Soil Acidity

- Nutrient availability reduced
- Toxic elements become more soluble
 - -Aluminum
 - Root pruning
 - Blocks sites of uptake on roots
 - Binds with P
 - Manganese



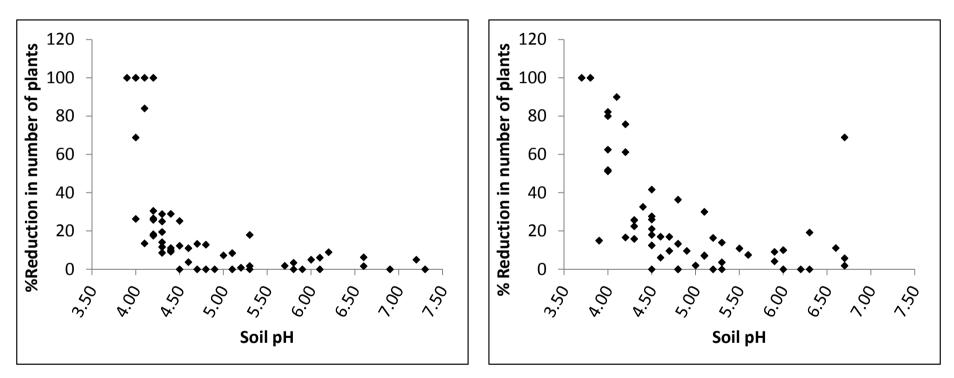
Plant grown under aluminum toxic conditions (acid soil conditions) compared with a normal plant the same age.

46% of the Oklahoma tested samples had a pH of <6.0 (PPI 2005).



Median Soil pH Values of OK Counties (all Ag. soils)

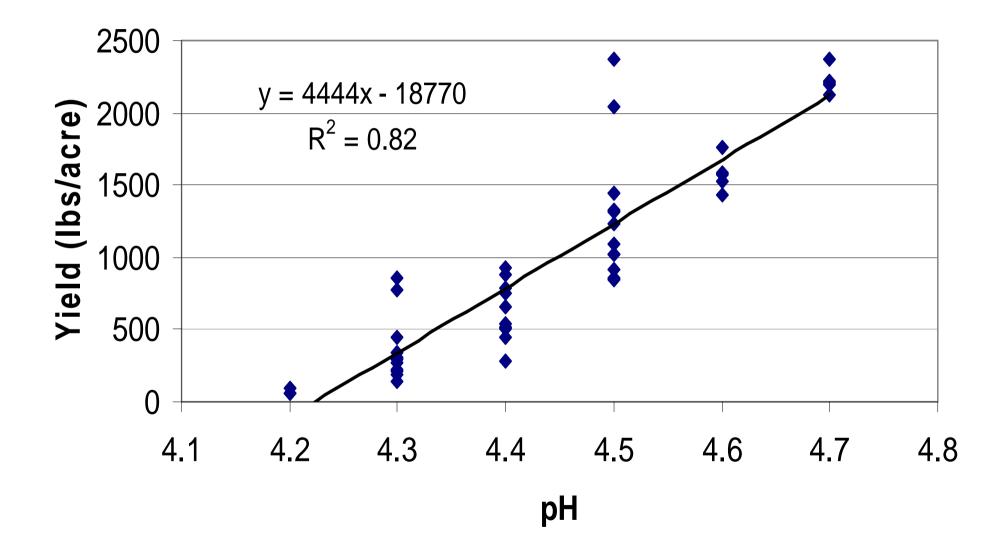
Emergence and Number of Heads at Harvest



Grain sorghum-Perkins, Lahoma, and Haskell, OK

Sunflower-Perkins, Lahoma, and Haskell, OK

Soil Acidity Reduces Forage Yield



Wheat

	Soil pH	Relative Yield (% of Max)	Actual Yield (bu/acre)	Loss (\$/acre)
	3.8	0	0	320.00
	4.1	30%	12	224.00
	4.5	60%	24	128.00
	5.0	85%	34	48.00
\rightarrow	5.5	95%	38	16.00
	6.0	100%	40	0.00

\$8.00 per bushel



Corn

	Soil pH	Relative Yield (% of Max)	Actual Yield (bu/acre)	Loss (\$/acre)
	4.3	32%	48	714.00
	4.6	74%	110	280.00
\rightarrow	4.8	80%	120	210.00
	5.3	88%	132	126.00
	5.6	96%	144	42.00
	6	100%	150	0.00

\$7.00 per bushel



Grain Sorghum

	Soil pH Relative Yield (% of Max)		Actual Yield (bu/acre)	Loss (\$/acre)
	4.0	40%	32	288.00
	4.3	50%	40	240.00
	4.6	60%	48	192.00
	4.9	70%	56	144.00
	5.1	80%	64	96.00
→	5.6	95%	76	24.00
	5.71	100%	80	0.00

\$6.00 per bushel



Sunflower

	Soil pH	Relative Yield (% of Max)	Actual Yield (lb/acre)	Loss (\$/acre)
	4.3	40%	580	261.00
	4.5	50%	725	217.50
	4.7	60%	870	174.00
	4.9	70%	1015	130.50
	5.1	80%	1160	87.00
\rightarrow	5.4	95%	1378	22.00
	5.5	100%	1450	0.00

\$0.30 per lb



Canola

	Soil pH Relative Yield (% of Max)		Actual Yield (bu/acre)	Loss (\$/acre)
	4.3	0%	0	480.00
	4.5	10%	4	432.00
	4.7	23%	9	372.00
	5.0	34%	14	312.00
	5.3	59%	24	192.00
\rightarrow	5.7	95%	38	24.00
	5.8	100%	40	0.00

\$12.00 per bushel





Soil pH 4.0

Soil pH 4.1

Soil pH 4.6



Soil pH 5.3

Soil pH 5.8



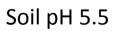
Soil pH 4.1

Soil pH 4.0

Soil pH 4.7



Soil pH 5.1



pH and Nutrients

strongly acid			medium acid	slightly acid	very slightly acid	very slightly alkaline	slightly alkaline	medium alkaiine	str	strongly alkaline		
-	-				n	trogen						
	_				p	hospho	orus			\$ 		
-	-				p	otassiu	m					
	-				S	ulphur					ni -	
-	-				Ci	alcium				-		
	-				m	agnes	ium					
	4		iron						-	-		
	-		mangar	iese						-	+	-
			boron							4		
			copper	& zinc						-		-
	_				п	olybde	enum					
1	4.5	5.0	5.5 6	i.0 I	6.5 7	7.0 7	.5 8	3.0 8	.5	9.0	9.5	1(



Phosphorus Deficiency

Phosphorus is **mobile** in the plant: Lower/Older Leaves Purpling of the leaf margins or base of stems. Symptoms similar too... Acidic soils



Iron Deficiency

Iron is **immobile** in the plant: Upper/Newer leaves Intervienal chlorosis, stripes narrower than zinc and extend full length of leaf Calcareous Soils pH >8



Zinc Deficiency

Zinc relatively **immobile** in plant: Upper/Newer leaves Purple margins, inward purple blotching, bleached bands on either side of midrib near base, intervienal chlorosis Acidic, Sandy soils, Calcareous pH>8



Manganese Deficiency

Manganese is relatively **immobile** in plant, but can move in xylem sap. Interveinal chlorosis, very similar to Fe, Mg, N. Severe deficiencies have brown specs and bronzing Limited at high pH, Calcareous soils.



Molybdenum Deficiency

Molybdenum is readily translocated; **whole plant** Related to N metabolism, yellowing, stunting, interveinal mottling Cupping of upper leaves Soil with low pH and high Fe and Al.

Lime required to neutralize the soil acidity produced by fertilizers if all ammonium-N is converted to nitrate-N.

Nitrogen source	Composition	Lime required (lb CaCO ₃ / lb N)					
Anhydrous ammonia	82-0-0	1.8					
Urea	46-0-0	1.8					
Ammonium nitrate	34-0-0	1.8					
Ammonium sulfate	21-0-0-24	5.4					
Monoammonium phosphate	10-52-0	5.4					
Diammonium phosphate	18-46-0	3.6					
Triple super phosphate	0-46-0	0.0					
Adapted from Havlin et al., 1999.							

Adapted from Haviin et al., 1999.